


# PHARMACOKINETICS

Laura M. Labay, Ph.D., F-ABFT, DABCC-TC



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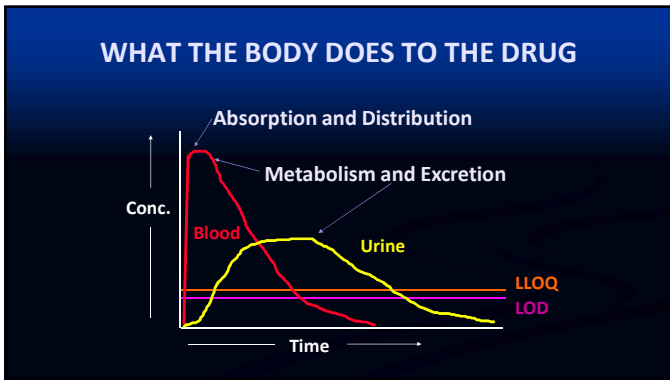
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### ABSORPTION

MOVEMENT OF DRUG FROM SITE OF ADMINISTRATION TO THE BLOODSTREAM



<http://www.4img.com/2017/02/researchers-use-light-to-track-drugs-in-blood-04/>

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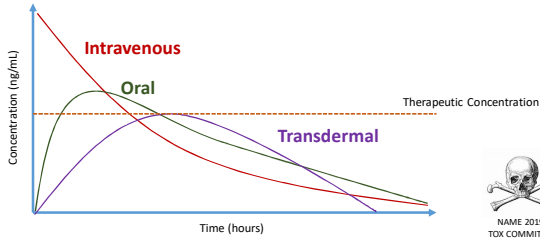
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## BIOAVAILABILITY - AUC

The proportion of a drug that enters the circulation when introduced into the body and so is able to have an active effect.




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## DETERMINANTS

- Physicochemical properties
- Formulation
- Route of administration




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## DEGREE OF IONIZATION

Degree of ionization  $\rightarrow$   $pK_a$

$pK_a \rightarrow$  the pH at which the ionized and unionized forms exist in equal concentrations

The more the drug is in its unionized form, the more likely it is to be lipid-soluble and transferred by passive diffusion through the membrane.




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## pK<sub>a</sub> RULES

### pK<sub>a</sub> RULES

pK<sub>a</sub> is defined as the pH where a drug exists as 50% ionized and 50% unionized

If pK<sub>a</sub> - pH = 0, then 50% of drug is ionized and 50% is unionized

Acidic groups become less ionized in an acidic environment  
Basic groups become less ionized in an alkaline environment




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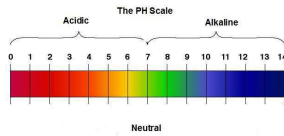
## QUANTIFYING THE DEGREE OF IONIZATION

For every unit by which pH is changed, the ratio of unionized to ionized molecules changes 10-fold

### EXAMPLE:

○ When the pH is 2 units less than the pK<sub>a</sub>, molecules of an acidic drug become 100x more un-ionized

pH = 3  
pK<sub>a</sub> = 5




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## PRACTICE QUESTIONS

- The pK<sub>a</sub> of sodium pentothal is 7.4 and the drug is acidic. If a patient is given sodium pentothal orally instead of by IV, will it put the patient to sleep? \_\_\_\_\_
- Absorption from the GI tract is determined by:
  - The pH of the medium and the pK<sub>a</sub> of the drug
  - The fraction of the drug that is non-ionized
  - Lipid solubility of the non-ionized drug
  - All of the above
  - None of the above
- Given a weak acid (pK<sub>a</sub> 8.0), the ratio of ionized to non-ionized form of the drug at pH 4.0 most closely approximates:
  - 2:1
  - 1:1
  - 1:1000
  - 10,000:1
  - None of the above
- Compared to a weak acid (pK<sub>a</sub> 8.0), another weak acid (pK<sub>a</sub> 5.3), other factors being equal, would be:
  - Absorbed from the GI tract at approximately the same rate
  - Absorbed from the GI tract more slowly
  - Absorbed from the GI tract more rapidly
  - Unabsorbed from the GI tract
  - More completely inactivated by gastric juice




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## PRACTICE QUESTIONS - ANSWERS

- 1) The  $pK_a$  of sodium pentothal is 7.4 and the drug is acidic. If a patient is given sodium pentothal orally instead of by IV, will it put the patient to sleep? **YES**
- 2) Absorption from the GI tract is determined by:
  - a) The pH of the medium and the  $pK_a$  of the drug
  - b) The fraction of the drug that is non-ionized
  - c) Lipid solubility of the non-ionized drug
  - d) **All of the above**
  - e) None of the above
- 3) Given a weak acid ( $pK_a$  8.0), the ratio of ionized to non-ionized form of the drug at pH 4.0 most closely approximates:
  - a) 2:1
  - b) 1:1
  - c) 1:1000
  - d) 10,000:1
  - e) **None of the above**
- 4) Compared to a weak acid ( $pK_a$  8.0), another weak acid ( $pK_a$  5.3), with all other factors being equal, would be:
  - a) Absorbed from the GI tract at approximately the same rate
  - b) **Absorbed from the GI tract more slowly**
  - c) Absorbed from the GI tract more rapidly
  - d) Unabsorbed from the GI tract
  - e) More completely inactivated by gastric juice




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## DISTRIBUTION

MOVEMENT OF A DRUG FROM ITS POINT OF ENTRY THROUGHOUT THE SYSTEMIC CIRCULATION AND INTO VARIOUS TISSUES

GOAL: FOR THE DRUG TO REACH ITS INTENDED SITE OF ACTION




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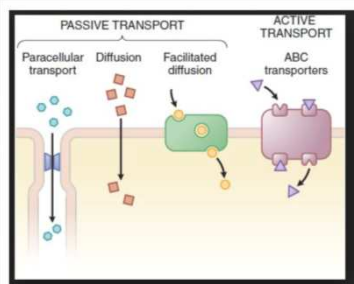
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## TRANSMEMBRANE MOVEMENT

- Molecular Size
- Drug Polarity
- Lipid Solubility
- Membrane Structure




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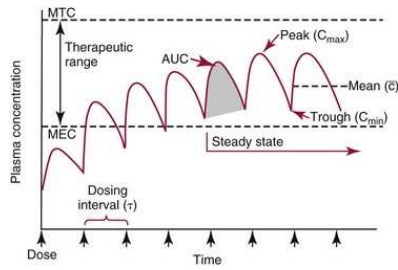
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## STEADY STATE CONCENTRATIONS




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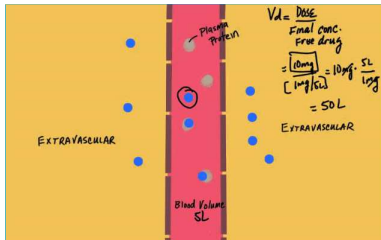
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## VOLUME OF DISTRIBUTION

A HYPOTHETICAL VOLUME INTO WHICH A DRUG IS DISTRIBUTED  
 IT IS A MEASURE OF HOW READILY DRUG DIFFUSES OUT OF THE PLASMA INTO TISSUES




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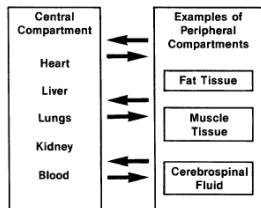
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## VOLUME OF DISTRIBUTION MEANING

1. Reflects the extent of drug distribution
2. In general,
  - Low  $V_d \rightarrow$  Drug confined to plasma
  - High  $V_d \rightarrow$  Drug equilibrates with tissues & extravascular fluids




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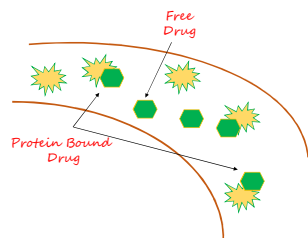
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## PROTEIN BINDING

Protein bound drug is:

- Not metabolized or excreted
- Inactive
- Confined to a specific tissue or site
- High plasma protein binding decreases  $V_d$




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## PROTEIN BINDING AND DISEASE

DRUG	UNBOUND %	UNBOUND % - DISEASE
Diazepam	2%	6% in liver disease
Furosemide	2%	6% in nephrotic syndrome
Phenytoin	9%	19% in renal disease
Triamterene	19%	40% in renal disease
Theophylline	35%	71% in liver disease
Digoxin	75%	82% in renal disease




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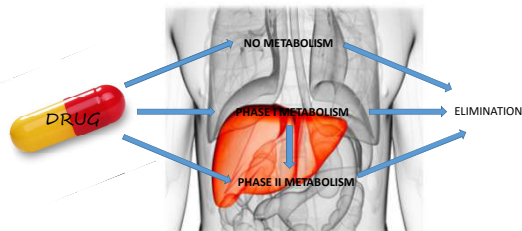
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## METABOLISM

BIOTRANSFORMATION OF PHARMACEUTICAL SUBSTANCES IN THE BODY




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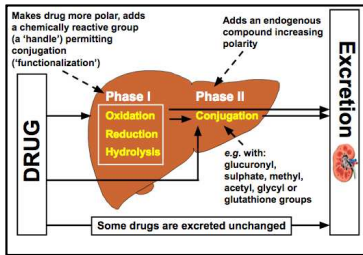
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## PHASE I AND II - OVERVIEW



<https://www.studydrive.com/notes/notes/vf/pharmacology/elementary-drug-metabolism-and-renal-excretion-of-drugs/563/7571438>




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## PHASE I METABOLISM

Chemical reactions – oxidation, reduction, hydrolysis  
Oxidation reactions are catalyzed by cytochrome p450 enzymes

Isoenzyme	Comments
CYP1A	Important for methylxanthines and paracetamol; induced by smoking
CYP2A	Limited number of substrates; significant interindividual variability
CYP2B	Limited number of substrates
CYP2C	CYP2C9 is an important isoform; CYP2C19 shows genetic polymorphism
CYP2D	Metabolises numerous drugs; CYP2D6 shows genetic polymorphism
CYP2E	Metabolises alcohol
CYP3A	Main isoform in liver and intestine; metabolises 50-60% of current drugs
CYP4	Metabolises fatty acids

Extreme caution should be taken if co-administration with a CYP3A4 inhibitor or inducer is unavoidable




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## CYTOCHROME p450 INTERACTIONS

### 1) COMPETITION

If the drugs are substrates for the same CYP isoform, the metabolism of each may be \_\_\_\_\_.

### 2) INHIBITION

In general, inhibition \_\_\_\_\_ plasma concentrations of substrate drugs, but \_\_\_\_\_ concentrations may be decreased.

### 3) INDUCTION

In general, induction \_\_\_\_\_ plasma concentrations of substrate drugs.




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### CYTOCHROME p450 INTERACTIONS

**COMPETITION**

If the drugs are substrates for the same CYP isoform, the metabolism of each may be **inhibited**.

**INHIBITION**

In general, inhibition **increases** plasma concentrations of substrate drugs, but **prodrug** concentrations may be decreased.

**INDUCTION**

In general, induction **decreases** plasma concentrations of substrate drugs.



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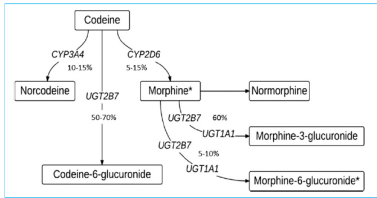
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### PHASE II METABOLISM

**"CONJUGATION REACTIONS"**

1. Substrates are coupled covalently to an endogenous molecule
2. Transferases catalyze the reaction
3. The resulting molecule is polar and able to be excreted in urine



<https://pubs.rsc.org/en/articlehtml/c6pp00017a>

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### METABOLIZER FORMS

- Rapid metabolizer
- Normal metabolizer
- Intermediate metabolizer
- Poor metabolizer



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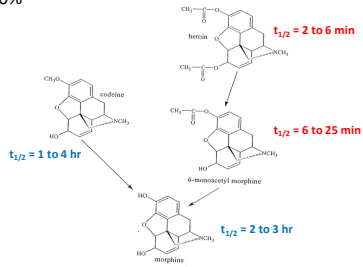
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### HALF-LIFE ( $t_{1/2}$ )

THE TIME IT TAKES FOR THE CONCENTRATION OF A DRUG IN PLASMA TO DECREASE BY 50%



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### FIRST ORDER ELIMINATION

Most drugs undergo first order elimination



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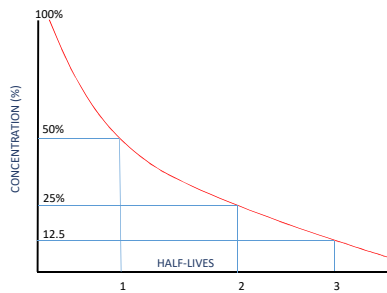
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### FIRST ORDER ELIMINATION



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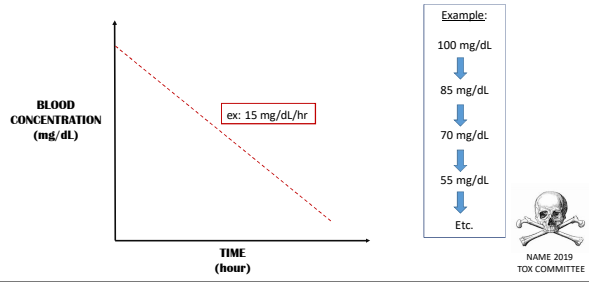
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## ZERO ORDER KINETICS

DEFINITION: A constant amount of drug is eliminated per unit time




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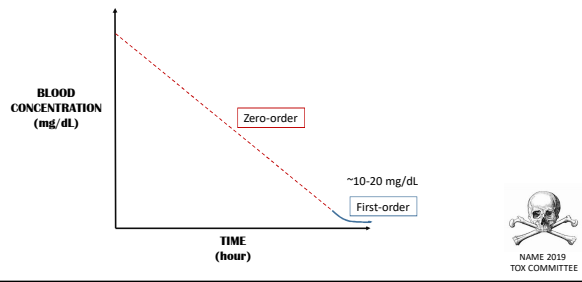
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## ZERO ORDER TO FIRST ORDER – ETOH EXAMPLE




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## FACILITATION OF ELIMINATION BASED ON pH

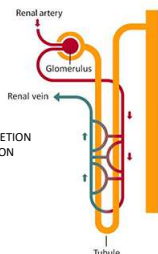
MATRIX	pH
URINE	4.8 to 8.0
BLOOD	7.4

URINE pH AFFECTS DRUG EXCRETION

EXAMPLES:

- METHAMPHETAMINE → ACIDIC URINE → INCREASES EXCRETION
- BARBITURATE → ACIDIC URINE → INCREASES REABSORPTION

The Nephron



<https://www.nidk.nih.gov/health-information/kidney-disease/kidneys-how-they-work>

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### ELIMINATION - QUESTION

A patient has presented to the Emergency Department with a Barbiturate overdose. Would you recommend that the urine should be made acidic or alkaline to facilitate excretion?



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### ELIMINATION - QUESTION

A patient has presented to the Emergency Department with a Barbiturate overdose. Would you recommend that the urine should be made acidic or alkaline to facilitate excretion?

I would want to make the urine alkaline.  
Why? This would cause the Barbiturate, a weakly acidic drug, to be ionized and not amenable to reabsorption.



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### POSTMORTEM PHARMACOKINETICS



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From OXYCONTIN® Package Insert for Oxycodone HCl Controlled-Release Tablets

Regimen	Dosage Form	AUC (ng•hr/mL)†	C <sub>max</sub> (ng/mL)	T <sub>max</sub> (hrs)	Trough Conc. (ng/mL)
Single Dose	10 mg OxyContin	100.7 [26.6]	10.6 [20.1]	2.7 [44.1]	n.a.
	20 mg OxyContin	207.5 [35.9]	21.4 [36.6]	3.2 [57.9]	n.a.
	40 mg OxyContin	423.1 [33.3]	39.3 [34.0]	3.1 [77.4]	n.a.
	80 mg OxyContin*	1085.5 [32.3]	98.5 [32.1]	2.1 [52.3]	n.a.
Multiple Dose	10 mg OxyContin Tablets q12h	103.6 [38.6]	15.1 [31.0]	3.2 [69.5]	7.2 [48.1]
	5 mg immediate-release q6h	99.0 [36.2]	15.5 [28.8]	1.6 [49.7]	7.4 [50.9]




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### ASSUMPTIONS

- The drug was ingested and all of it was absorbed at the time of specimen collection.
- The drug was ingested only one time (e.g., a single dose).
- The drug concentration represents the peak (C<sub>max</sub>) blood-drug concentration.
- The drug concentration accurately reflects the circulating blood-drug concentration in the antemortem state in that it has been unaffected by such influences such as postmortem redistribution.
- The presence of metabolites and/or the drug's presence in other matrix types are not represented.
- The volume of distribution (V<sub>d</sub>) and the blood to plasma drug ratio (b/p) of the drug is known.


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### TRUE or FALSE

1. The conjugation of morphine with glucuronide acid decreases the narcotic activity of the drug hastens excretion.
2. The metabolism of diazepam to nordiazepam is an example of an oxidative (Phase I) reaction.
3. The ionized form of a drug is the form which readily crosses the membrane.
4. If a weakly basic drug is given by IV, it will not be found in the stomach.
5. A weak acid would demonstrate an increased renal clearance if the urine was alkalinized.
6. A weakly acidic drug is equally absorbed from the stomach and small intestine.
7. Normally, the blood has a pH of 7.4




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### TRUE or FALSE

1. The conjugation of morphine with glucuronide acid decreases the narcotic activity of the drug hastens excretion. **TRUE**
2. The metabolism of diazepam to nordiazepam is an example of an oxidative (Phase I) reaction. **TRUE**
3. The ionized form of a drug is the form which readily crosses the membrane. **FALSE**
4. If a weakly basic drug is given by IV, it will not be found in the stomach. **TRUE**
5. A weak acid would demonstrate an increased renal clearance if the urine was alkalinized. **TRUE**
6. A weakly acidic drug is equally absorbed from the stomach and small intestine. **FALSE**
7. Normally, the blood has a pH of 7.4 **TRUE**



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Pharmacokinetics is concerned with the rate at which drugs enter the body, distribute within it, and then leave. Includes metabolism.



The  $V_d$  of a drug is the volume in which it would need to distribute so that the concentration in the body is equal to that of the blood.

Metabolic reactions tend to make a drug progressively more water soluble so it can be eliminated in the urine.

Most drugs are eliminated following first-order processes.



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### QUESTIONS?



Laura.Labay@nmslabs.com



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**Genetic Polymorphisms**  
Jirair Gevorkyan, Ph.D., F-ABFT



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**Objectives**

1. Understand the basis and application of genetic information in relation to drug response and toxicity.
2. To distinguish genetics and enzymes behind strong and weak metabolic responders.
3. Understand how genetic polymorphisms can be applied to postmortem toxicology and death investigations.



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**Outline**

- Introductory Concepts
  - Central Dogma
  - Mutations
  - Structure and Function
- Polymorphisms
  - Consequences in Pharmacology
  - Cytochrome P450s
- Application to Toxicology
  - Relevant Polymorphisms and Drugs
  - Case Examples



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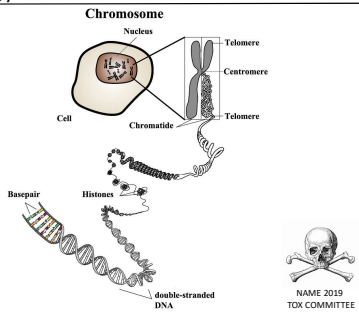
### Central Dogma of Biology

**Hereditary information**

- Humans Chromosome: 46
- Dogs: 78
- Cats: 38
- Fruit Flies: 8

**Consistent Differences**

- Between species
- Between subpopulations
- Between individuals




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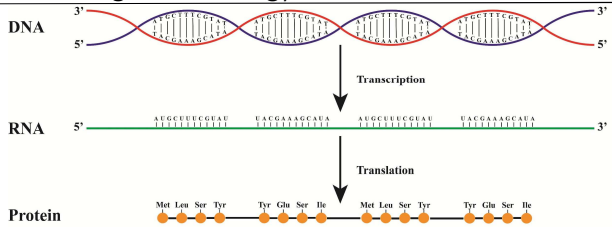
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### Central Dogma of Biology



But also...  
 Effects of ribozymes, proteins, metabolites produce bidirectionality  
 Post transcriptional/translational modifications, somatic epitypes, epigenetics




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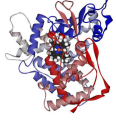
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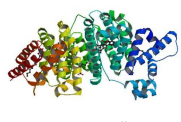
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### Central Dogma of Biology

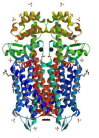
- Structure -> Function
- Genotype -> Phenotype



CYP2C9



Human Serum Albumin



μ-opioid receptor




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### Mutations

**Single Nucleotide Polymorphisms**

3' ACGCCTTGACGA**G**GCTTAC 5'  
 5' TCGGAACTGCT**B**CGAATG 3'

3' ACGCCTTGACGA**A**GCTTAC 5'  
 5' TCGGAACTGCT**T**CGAATG 3'

**Insertion/Deletion**

3' ACGCCTTGACGA**AG**GCTTAC 5'  
 5' TCGGAACTGCT**AG**CGAATG 3'

3' ACGCCTTGACGA**AGCAGCAGCAGC**GCTTAC 5'  
 5' TCGGAACTGCT**AGCAGCAGCAGCAGC**CGAATG 3'

**Short Tandem Repeat**

3' ACGCCTTGACGA**AGCAGCAGC**GCTTAC 5'  
 5' TCGGAACTGCT**AGCAGCAGC**GAATG 3'

3' ACGCCTTGACGA**AGCAGCAGCAGCAGC**GCTTAC 5'  
 5' TCGGAACTGCT**AGCAGCAGCAGCAGCAGC**CGAATG 3'

		Second letter						
		U	C	A	G			
U	UUU	Pha	UCU	Ser	UAU	Tyr	UGU	Cys
	UUA	Leu	UCC	Ser	UAC	Tyr	UGC	Cys
	UUG	Leu	UCA	Ser	UAA	Stop	UGA	Stop
C	CUU	Leu	CCU	Pro	CAU	His	CCU	His
	CUC	Leu	CCC	Pro	CAC	His	CCC	His
	CUA	Leu	CCA	Pro	CAA	Gln	CGA	Arg
A	AUU	Ile	ACU	Thr	AUU	Asn	AGU	Ser
	AUC	Ile	ACC	Thr	AUA	Asn	AGA	Arg
	AUA	Met	ACA	Thr	AAA	Lys	AGG	Arg
G	GUU	Val	GCU	Ala	GAU	Asp	GGU	Gly
	GUC	Val	GCC	Ala	GAC	Asp	GGC	Gly
	GUA	Val	GCA	Ala	GAA	Glu	GGG	Gly




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### Mutations, Structure and Function

DNA: TCATATGCACCCGT  
 Peptide: S Y A P R

**Mutation**

DNA: TCATATGCAC**AG**CGT  
 Peptide: S Y A **E** R

**Silent Substitution**

DNA: TCATATGCAC**GC**CGT  
 Peptide: S Y A **R** R

**Dissimilar Substitution**

DNA: TCATATGCAT**AT**CGT  
 Peptide: S Y A **Y** R

**Insertion**

DNA: TCATATGGA**AGAGC**CCCGT  
 Peptide: S Y A **S G** P R




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### Genetic Polymorphisms

**Inter-individual genetic variability**

Person 1 Chromosome 5  
 Copy 1: 3'-TGACGA**T**GCTTAC 5'  
 5'-ACTGCT**A**CGAATG 3'

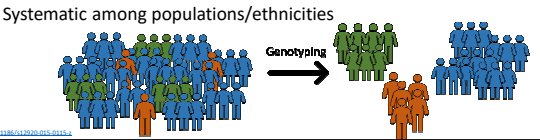
Person 2 Chromosome 5  
 Copy 1: 3'-TGACGA**G**GCTTAC 5'  
 5'-ACTGCT**C**CGAATG 3'

Person 3 Chromosome 5  
 Copy 1: 3'-TGACGA**C**GCTTAC 5'  
 5'-ACTGCT**G**CGAATG 3'

Copy 2: 3'-TGACGA**A**GCTTAC 5'  
 5'-ACTGCT**T**CGAATG 3'

Copy 2: 3'-TGACGA**C**GCTTAC 5'  
 5'-ACTGCT**G**CGAATG 3'

Copy 2: 3'-TGACGA**G**GCTTAC 5'  
 5'-ACTGCT**C**CGAATG 3'




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## Genetic Polymorphisms

### Pharmacogenetics

- Individual gene-drug interactions, usually one or two genes that have dominant effect on a drug response

### Pharmacogenomics

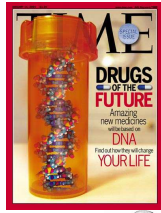
- Genomic influence on drug response, often using high-throughput data, sequencing, SNP chip, expression, proteomics

### Pharmacodynamic

- Receptors, ion channels, immune molecules

### Pharmacokinetic

- Transporters, **metabolic enzymes**, plasma protein binding




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## Metabolizer Forms

- Normal metabolizer



- Intermediate metabolizer



- Poor metabolizer



- Ultrarapid metabolizer



Parent drug and metabolite activity are important

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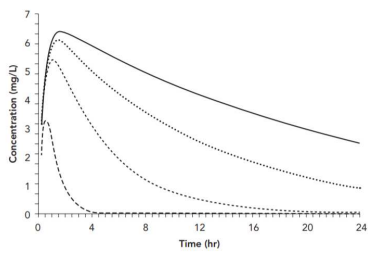
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## Metabolizer Forms



— Poor metabolizer      ··· Intermediate metabolizer  
 - - - Ultrarapid Metabolizer      ···· Extensive (normal) metabolizer




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## Metabolizer Forms

Analyte	Poor Metabolizer	Ultrarapid Metabolizer
Prodrug, <b>Active Metabolite</b>	poor efficacy, accumulation of prodrug	good efficacy, rapid effect
<b>Active Drug</b> , Inactive Metabolite	good efficacy, but accumulation may have adverse effects	poor efficacy, need a greater dose or slow release formula

(A) Normal Metabolizer

(B) Poor Metabolizer

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## Practice Question

- 1) A mutation is defined as any change in a DNA sequence away from normal. A polymorphism is a DNA sequence variation that is common in the population. TRUE or FALSE
- 2) A poor metabolizer may have one of the following genetic predispositions:
  - a) Two functioning alleles related to drug metabolism
  - b) One functioning and one non-functioning alleles related to drug metabolism
  - c) Two non-functioning alleles related to drug metabolism
  - d) More than two functioning alleles related to drug metabolism
- 3) An ultra rapid metabolizer may have one of the following genetic predispositions:
  - a) Two functioning alleles related to drug metabolism
  - b) One functioning and one non-functioning alleles related to drug metabolism
  - c) Two non-functioning alleles related to drug metabolism
  - d) More than two functioning alleles related to drug metabolism
- 4) Pharmacogenetics is the study of
  - a) study of how people respond differently to drug therapy based upon their genetic makeup or genes
  - b) how genes affect a person's response to drugs
  - c) the science of drugs
  - d) the study of genes and their functions

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## Practice Question

- 1) A mutation is defined as any change in a DNA sequence away from normal. A polymorphism is a DNA sequence variation that is common in the population. **TRUE** or FALSE
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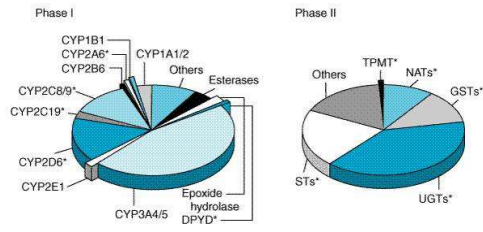
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## Metabolic Enzymes



Other important genetic polymorphisms outside of metabolic enzymes (VKORC1, ABCE)



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## Genetic Polymorphisms

Table 1. Pharmacogenetics of Phase I Drug Metabolism.<sup>a</sup>

Drug-Metabolizing Enzyme	Frequency of Variant Poor-Metabolism Phenotype	Representative Drugs Metabolized	Effect of Polymorphism
Cytochrome P-450 2D6 (CYP2D6)	6.8% in Sweden 1% in China <sup>19</sup>	Debrisoquin <sup>19</sup> Sparine <sup>21</sup> Nortriptyline <sup>23</sup> Codeine <sup>27,28</sup>	Enhanced drug effect Enhanced drug effect Enhanced drug effect Decreased drug effect
Cytochrome P-450 2C9 (CYP2C9)	Approximately 3% in England <sup>29</sup> (those homozygous for the *2 and *3 alleles)	Warfarin <sup>30</sup> Phenytoin <sup>31,32</sup>	Enhanced drug effect <sup>29,32</sup>
Cytochrome P-450 2C19 (CYP2C19)	2.7% among white Americans <sup>33</sup> 3.3% in Sweden 14.6% in China <sup>37</sup> 18% in Japan <sup>33</sup>	Omeprazole <sup>34,35</sup>	Enhanced drug effect <sup>33,37</sup>
Dihydropyrimidine dehydrogenase	Approximately 1% of population is heterozygous <sup>36</sup>	Fluorouracil <sup>36,40</sup>	Enhanced drug effect <sup>36,40</sup>
Butyrylcholinesterase (pseudocholinesterase)	Approximately 1 in 3500 Europeans <sup>41</sup>	Succinylcholine <sup>41</sup>	Enhanced drug effect <sup>41</sup>

<sup>a</sup> Examples of genetically polymorphic phase I enzymes are listed that catalyze drug metabolism, including selected examples of drugs that have clinically relevant variations in their effects.



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## Cytochrome P450 (CYP450)

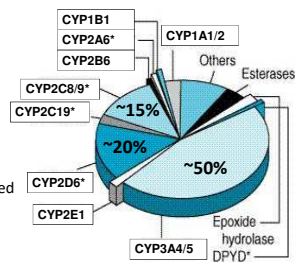
Critical for metabolism of drugs

- Oxidize drugs
- Making drugs more water-soluble

Clinically relevant if inactivated

- Inhibition (grapefruit)
- Downregulation
- Drug-Drug Interaction
- CYP3A4 covers ~50 %
  - Most consequential if inactivated

- CYP2C9\*2A
- Superfamily
  - Family
  - Subfamily
  - Isozyme
  - Allele
  - Suballele



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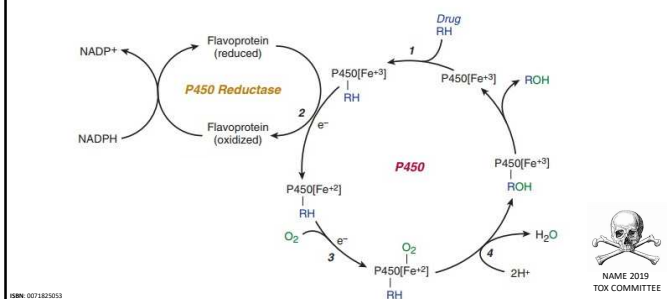
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### CYP450 Mechanism




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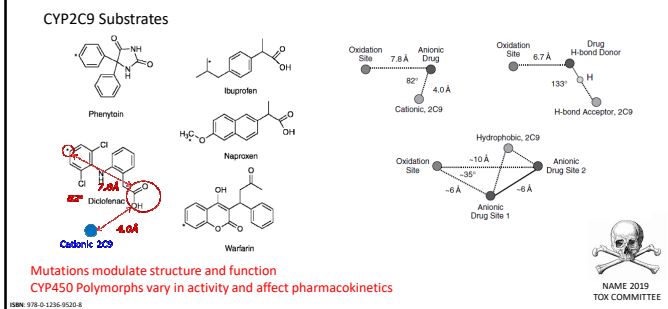
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### CYP450 Structure and Function




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### CYP450 Polymorphisms

TABLE 1  
Importance of Polymorphic CYP for the Metabolism of Drugs and Carcinogens

Enzyme	Substrate	Polymorphism frequency	Functional effect	Non-Importance pharmacokinetic status
CYP1A1	Carcinogen	Relatively high	Upstream	No important functional variant alleles
CYP1A2	Drugs, carcinogen	High	Rare	CYP1A2*10
CYP1B1	Carcinogen, estrogen	Rare and allele, frequent missense mutations	All but seven haplotypes with variable activity	CYP1B1*3
CYP2A6	Nicotine, drug, carcinogen	High in Caucasians, low frequent in Caucasians	Important for nicotine metabolism	CYP2A6*10, CYP2A6*9, CYP2A6*7, CYP2A6*6, CYP2A6*5
CYP2B6	Drugs	High	Reduced drug metabolism	CYP2B6*5
CYP2C8	Beer drugs	High	Reduced drug metabolism	CYP2C8*3
CYP2C9	Drugs	Relatively rare in Caucasians	Very significant	CYP2C9*2, CYP2C9*3
CYP2C19	Drugs	High	Very significant	CYP2C19*2, CYP2C19*17, CYP2C19*11
CYP2D6	Drugs	Very high	Very significant	CYP2D6*20, CYP2D6*4, CYP2D6*10, CYP2D6*17
CYP2E1	Carcinogen, vitamins, free drugs	Low	No significant ones have been reported	No important functional variant alleles
CYP3A4	Drugs, carcinogen	Low	No or small	CYP3A4*28
CYP3A5	Drugs, carcinogen	High	Negligible	CYP3A5*1, CYP3A5*2
CYP3A7	Drugs, carcinogen	Low	None	CYP3A7*2

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## Practice Question

- Cytochrome P450s catalyze
  - Hydrolysis
  - Redox reactions
  - Dealkylation
  - Deamination
  - All of the above
- Most clinically relevant inhibitory interactions are the consequence of inactivation of which enzyme?
  - CYP1A2
  - CYP2D6
  - CYP2C9
  - CYP2C19
  - CYP3A4
- What type of metabolizer form would be expected for an individual with multiple copies of non-functioning a cytochrome P450? **Poor metabolizer**
- Which of the two following CYP450 isozymes exhibit significant genetic polymorphism?
  - CYP1A1
  - CYP2B6
  - CYP2D6**
  - CYP2C19**
  - CYP2E1




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## Application to Toxicology

- Pharmacogenetics based dose and clinical outcome
- Phenotype consideration in death investigations

Enzyme	Drug Metabolized
CYP2C19	Amitriptyline, Imipramine, Diazepam, Citalopram, Carisoprodol, Clopidogrel, Desipramine, Omeprazole, <b>Phenytoin</b>
CYP2D6	<b>Amphetamines</b> , <b>Codeine</b> , Oxycodone, Hydrocodone, Methadone, Tramadol, Dextromethorphan, Metoclopramide, Desipramine, Metoprolol, Amitriptyline, Duloxetine, Fluoxetine, Haloperidol, Risperidone, Thioridazine
CYP2C9	<b>NSAIDs</b> , Valproic acid, <b>Warfarin</b> , <b>Phenytoin</b> , Glipizide, Ibuprofen, Celecoxib, Fluvastatin
CYP3A4	<b>Benzodiazepines</b> , Fentanyl, Methadone, Buprenorphine, Cocaine, Zolpidem, <b>Antibiotics</b> , <b>Calcium Channel Blockers</b> , <b>Statins</b> , <b>Steroids</b>
CYP2E1	Ethanol, Acetaminophen




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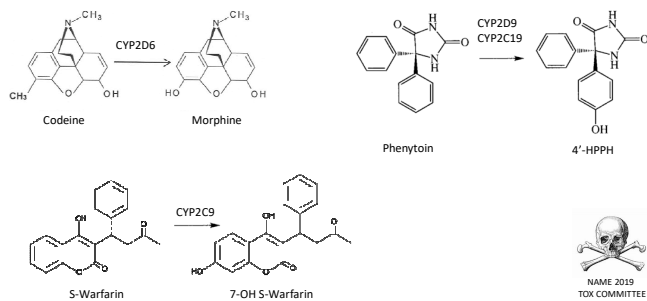
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## Application to Toxicology




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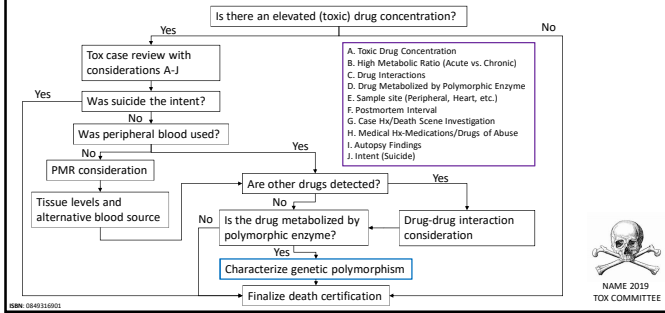
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### Application to Death Investigation




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### Methodology

- Collection and preparation of samples
- Extraction by ionic resins or commercial kits (QIAGEN)
- Separation of DNA by gel electrophoresis
- Genotyping
  - Real-Time PCR System
  - Capillary Electrophoresis
  - MALFI-TOF




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### Case 1 (2D6 and 2C19 Poor Metabolizer)

- 46 year old Caucasian male diagnosed with depression
- Desipramine 50-250 mg daily over 1 month treatment as part of a safety and efficacy study
- Chest tightness occurring about 2 hours after desipramine dose, always while he was at rest
- Hospitalized when chest pain lasted 1 hour, ECG consistent with acute anterior wall myocardial ischemia
- Plasma desipramine concentration at 764 ng/mL
- Patient was studied with the drug-metabolizing probes, debrisoquin, mephenytoin, and dapsone
- Found to be poor metabolizer for both CYP2D6 and CYP2C19




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### Case 5 (2C9 Poor metabolizer)

- Elderly woman receiving 2.5 mg warfarin daily after a pacemaker one year previously
- Given celecoxib for joint pain 4 weeks prior
- Rx also included digoxin, ranitidine, and atorvastatin
- Developed ecchymoses with a decline in hemoglobin concentration from 160 to 85 g/L over one week
- Was found to be heterozygous for CYP2C9\*2 and \*3
- Metabolism of warfarin was perturbed and led to bleeding in the presence of variant CYP2C9 genes by coadministration of celecoxib



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### Limitations

- Genetic variation is only part of the puzzle, context is important
- Does not account for post translational modification
- Does not account for expressional modification
- Studies are primarily clinical, limited in postmortem
- Flat application of ethnic/race to phenotype is strongly discouraged
- Insightful information requires genetic testing
- Must consider postmortem interval, redistribution, age, environmental factors, drug interactions, etc




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### Summary

- Genetic polymorphisms are inter-individual variations in genetic code that are also systematic among populations/ethnicities
- Genetic Polymorphisms in metabolic enzymes like CYP2D6 and CYP2C19 contributes to inter-individual differences in drug response
- Examples of drug toxicities that can be predicted by P450 polymorphism include those exerted by codeine, tramadol, warfarin, etc
- Genetic polymorphisms can assist in the interpretation of drug concentrations in postmortem toxicology and drug death certification




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### Questions?



"Here's my  
sequence..."

*New Yorker, 2000*

Jirair Gevorkyan  
jgevorkyan@outlook.com



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## Drug Interactions

Luigino Apollonio, Ph.D.

NAME 2019 Annual Meeting – 20 October 2019



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### Outline

- Background
- Types and mechanisms of drug interactions
- Consequences
- Factors influencing interactions
- Avoiding interactions
- Reference information
- Case studies
- Final thoughts



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### Objectives

1. To identify the types and mechanisms of drug interactions for consideration in postmortem toxicology and death investigation
2. To identify factors influencing drug interactions, and potential consequences of interactions
3. To become more familiar with key resources on drug interactions, and to explore polypharmacy in case studies




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### Background

- Interaction between two or more drugs (or something else!) that prevents them from acting as expected
  - Drug-drug, food/beverage-drug, condition-drug
- A drug interaction may:
  - Affect the total population
  - Affect a particular subset of the population




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### Background

- Contraindicated
  - High risk of severe interaction – do not use
- Serious
  - Potential for a serious interaction – may require regular monitoring or alternative medication(s)
- Significant
  - Potential for an interaction – may require monitoring
- Minor
  - Interaction may not be significant or may be unlikely




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**Background**

- Polypharmacy
  - Use of multiple medications
    - Benefit-Risk Assessment
  - Associated risks across general and specialist care
- DIs estimated to cause over 2% of annual hospitalizations in the USA (Carpenter et al 2019)
  - Bethi et al (2018): From a total of 433 prescriptions (46%) had one or more potential DDIs (range of 1-13 DIs per prescription)
    - Older patients and those prescribed >6 drugs were at 'major risk'
  - Gujjarlamudi (2016): Prevalence of 'inappropriate' medication use in the elderly ranges from 11.5-62.5% (citing Guaraldo et al 2011)
    - Risk of DI estimates at 13% for two drugs, 58% for five drugs, and 82% for seven or more drugs (citing Fulton and Allen, 2005)




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**Background**

- Pharmacokinetic interactions
  - Increase or decrease in:
    - Absorption
    - Distribution
    - Metabolism
    - Elimination
- Additive
- Synergistic
- Antagonistic
- Pharmacodynamic interactions
  - Homodynamic (same receptor)
    - (pure/partial agonists; competitive/noncompetitive/uncompetitive antagonists)
  - Heterodynamic (different receptors)




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**Mechanisms**

- Absorption mechanisms
  - Change in intestinal blood flow
  - Change in metabolism in the intestine
  - Change in gastric emptying/intestinal motility
  - Change in gastric acidity
  - Change in intestinal flora
  - Change in solubility
- Distribution mechanisms
  - Effects of body changes – lean muscle, increased fat
  - Changes in protein binding




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### Mechanisms

- Metabolism mechanisms
  - Cytochrome p450 effects
    - Leads to changes in the concentration and effect of a drug or its metabolites
    - Enzyme induction
    - Enzyme inhibition
- Elimination mechanisms
  - Renal effects – decrease in renal blood flow, GFR, tubular secretion
  - Urine pH – weak acids or bases in the urine
  - Biliary excretion – reabsorption in the intestine (enterohepatic recirculation)



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### Consequences

- Increase or decrease in the beneficial or adverse effects of drugs
  - Reduction in desired effects
    - failure of therapy
    - may need increased dosage
  - Increase in adverse effects
    - Increase in frequency or severity of adverse effects
- May be unpredictable
- May lead to medication non-compliance



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### Practice Questions

- 1) Drug interactions may be categorized (not least) to pharmacokinetic interactions and pharmacodynamic interactions? TRUE or FALSE
- 2) The risk of an adverse drug event has been estimated at what percent (%) for a person taking five drugs?
  - a) 13%
  - b) 28%
  - c) 58%
  - d) 82%
  - e) 85%
- 3) Consequences of drug interactions include which of the following?
  - a) Reduction in desired effects
  - b) Failure of therapy
  - c) Increase in adverse effects
  - d) Increase in frequency or severity of side effects
  - e) All the above



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### Practice Questions

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### Factors

- Genetics
- Physiology/pathophysiology/disease (hepatic, renal)
- Age
- Lifestyle
- Dosing regimen
- Duration of therapy
- Time(s) of administration
- Degree of, and agents in, polypharmacy
- Therapeutic index



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### How are patients to avoid DIs?

- Ask HCPs about interactions
- Keep a list of current medications
- Inform HCPs of medications, and include OTCs, foods, vitamins, supplements, herbals/traditional medications
- Inform HCPs when medications are stopped/started
- Inform HCPs about lifestyle changes
- Understand why one is taking each medication
- Be aware of side effects
- Try to eliminate unnecessary meds – ask to simplify meds or discontinue if possible
- Take only as prescribed



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### Tools: Prescribing Information

- Warnings and Precautions (includes Black Box Warnings)
- Dosage and Administration
- Contraindications
- Adverse Reactions
- Drug interactions
- Use in specific populations
- Examples: Warfarin, Sertraline, Mycophenolic acid, Metoprolol




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### Warfarin PI

- **Black Box Warning**
  - Drugs, dietary changes, and other factors affect INR levels achieved with coumadin therapy.
- **Drug Interactions**
  - Concomitant use of drugs that increase bleeding risk, antibiotics, antifungals, botanical (herbal) products, and inhibitors and inducers of CYP2C9, 1A2, or 3A4.
  - Consult labeling of all concurrently used drugs for complete information about interactions with [warfarin] or increased risks for bleeding.




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### Warfarin PI

- **Drug Interactions**
  - Drugs may interact with [warfarin] through pharmacodynamic or pharmacokinetic mechanisms.
  - Pharmacodynamic mechanisms for drug interactions with [warfarin] are synergism (impaired hemostasis, reduced clotting factor synthesis), competitive antagonism (vitamin K), and alteration of the physiologic control loop for vitamin K metabolism (hereditary resistance).
  - Pharmacokinetic mechanisms for drug interactions with [warfarin] are mainly enzyme induction, enzyme inhibition, and reduced plasma protein binding.
  - It is important to note that some drugs may interact by more than one mechanism.




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### Sertraline PI

<b>Drugs that Interfere with Hemostasis (antiplatelet agents and anticoagulants)</b>	
<i>Clinical Impact</i>	The concurrent use of an antiplatelet agent or anticoagulant with ZOLOFT may potentiate the risk of bleeding.
<i>Intervention</i>	Inform patients of the increased risk of bleeding associated with the concomitant use of ZOLOFT and antiplatelet agents and anticoagulants. For patients taking warfarin, carefully monitor the international normalized ratio (See <i>Warnings and Precautions</i> (1.3)).
<i>Examples</i>	aspirin, clopidogrel, heparin, warfarin
<b>Drugs Highly Bound to Plasma Protein</b>	
<i>Clinical Impact</i>	ZOLOFT is highly bound to plasma protein. The concomitant use of ZOLOFT with another drug that is highly bound to plasma protein may increase free concentrations of ZOLOFT or other tightly-bound drugs in plasma (See <i>Clinical Pharmacology</i> (12.3)).
<i>Intervention</i>	Monitor for adverse reactions and reduce dosage of ZOLOFT or other protein-bound drugs as warranted.
<i>Examples</i>	warfarin
<b>Drugs Metabolized by CYP2D6</b>	
<i>Clinical Impact</i>	ZOLOFT is a CYP2D6 inhibitor (See <i>Clinical Pharmacology</i> (12.3)). The concomitant use of ZOLOFT with a CYP2D6 substrate may increase the exposure of the CYP2D6 substrate.
<i>Intervention</i>	Decrease the dosage of a CYP2D6 substrate if needed with concomitant ZOLOFT use. Conversely, an increase in dosage of a CYP2D6 substrate may be needed if ZOLOFT is discontinued.
<i>Examples</i>	propafenone, flecainide, sotalolol, desipramine, dexchlorpheniramine, metoprolol, nebivolol, perphenazine, thioridazine, tolterodine, venlafaxine
<b>Phenytoin</b>	
<i>Clinical Impact</i>	Phenytoin is a narrow therapeutic index drug. ZOLOFT may increase phenytoin concentrations.
<i>Intervention</i>	Monitor phenytoin levels when initiating or titrating ZOLOFT. Reduce phenytoin dosage if needed.
<i>Examples</i>	phenytoin, fosphenytoin



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### Mycophenolic acid PI

- **Warnings and Precautions**
  - A variety of drugs have potential to alter systemic MPA exposure when co-administered with [MPA].
  - Therefore, determination of MPA concentrations in plasma before and after making any changes to immunosuppressive therapy, or when adding or discontinuing concomitant medications, may be appropriate to ensure MPA concentrations remain stable.
- **Drug Interactions**
  - See FPI for drugs that may interfere with systemic exposure and reduce [MPA] efficacy: antacids with magnesium or aluminum hydroxide, proton pump inhibitors, drugs that interfere with enterohepatic recirculation, telmisartan, calcium-free phosphate binders.



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### Mycophenolic acid PI

<b>Antacids with Magnesium or Aluminum Hydroxide</b>	
<i>Clinical Impact</i>	Concomitant use with an antacid containing magnesium or aluminum hydroxide decreases MPA systemic exposure (see <i>Clinical Pharmacology</i> (12.3)), which may reduce CELLCEPT efficacy.
<i>Prevention or Management</i>	Administer magnesium or aluminum hydroxide containing antacids at least 2h after CELLCEPT administration.
<b>Proton Pump Inhibitors (PPIs)</b>	
<i>Clinical Impact</i>	Concomitant use with PPIs decreases MPA systemic exposure (see <i>Clinical Pharmacology</i> (12.3)), which may reduce CELLCEPT efficacy.
<i>Prevention or Management</i>	Monitor patients for alterations in efficacy when PPIs are co-administered with CELLCEPT.
<i>Examples</i>	Lansoprazole, pantoprazole



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## Mycophenolic acid PI

<b>Clinical Impact</b>	Concomitant use with cell-surface kinase phosphatases decreases MPAs systemic exposure (see Clinical Pharmacology (12.2)), which may reduce CELLCEPT efficacy.
<b>Prevention or Management</b>	Administer cell-surface kinase phosphatase inhibitors at least 2 hours after CELLCEPT.
<b>Examples</b>	Sevelamer

**Table 6. Drug Interactions with CELLCEPT that Affect Other Drugs**


Drug that Undergoes Metabolism	Underlying Interaction
<b>Clinical Impact</b>	When administered with CELLCEPT, an anticholinergic (MPAG) may compete with drug elimination by renal tubular secretion, which may increase plasma concentrations and/or adverse reactions associated with these drugs.
<b>Prevention or Management</b>	Monitor for drug-related adverse reactions in patients with renal impairment.
<b>Examples</b>	Acetaminophen, gabapentin, propofol, valacyclovir, valproic acid

<b>Clinical Impact</b>	Concomitant use with CELLCEPT decreased the systemic exposure to efavirenz, but did not affect the systemic exposure to efavirenz/tenofovir (see Clinical Pharmacology (12.2)), which may result in reduced combination oral contraceptive effectiveness.
<b>Prevention or Management</b>	Use additional barrier contraceptive methods.

<b>Clinical Impact</b>	Concomitant use with PPIs decreases MPAs systemic exposure (see Clinical Pharmacology (12.2)), which may reduce CELLCEPT efficacy.
<b>Prevention or Management</b>	Monitor patients for decreases in efficacy when PPIs are co-administered with CELLCEPT.
<b>Examples</b>	Lansoprazole, pantoprazole

<b>Clinical Impact</b>	Concomitant use with drugs that directly interact with anticholinergic receptors, or indirectly interact with anticholinergic receptors (e.g., indirectly interact with anticholinergic receptors by slowing the gastrointestinal motility) may decrease MPAs systemic exposure (see Clinical Pharmacology (12.2)), which may reduce CELLCEPT efficacy.
<b>Prevention or Management</b>	Monitor patients for decreases in efficacy or CELLCEPT-related adverse reactions when these drugs are co-administered with CELLCEPT.
<b>Examples</b>	Tricyclic antidepressants, skeletal muscle relaxants, anticholinergics, antipsychotics, as well as muscarinic antagonists, benzodiazepines, and other drugs that slow GI motility

<b>Clinical Impact</b>	Concomitant use with drugs reducing gastrointestinal motility decreases MPAs systemic exposure, potentially reducing CELLCEPT efficacy, which may reduce combination oral contraceptive effectiveness.
<b>Prevention or Management</b>	Monitor patients for decreases in efficacy or CELLCEPT-related adverse reactions when these drugs are co-administered with CELLCEPT.
<b>Examples</b>	Tricyclic antidepressants, anticholinergics, skeletal muscle relaxants, antipsychotics, and other drugs that slow GI motility



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## Metoprolol PI

- **Warnings and Precautions**
  - Bronchospastic Disease: Avoid beta blockers [use lowest dose, or consider concomitant beta2-agonists]
  - Pheochromocytoma: First initiate therapy with an alpha blocker [due to observed paradoxical increase in BP]
- **Drug Interactions**
  - Catecholamine-depleting drugs may have an additive effect when given with beta-blocking agents.
  - Patients may be unresponsive to the usual doses of epinephrine used to treat allergic reaction.
  - CYP2D6 inhibitors are likely to increase metoprolol concentration.
  - Concomitant use of glycosides, clonidine, and diltiazem and verapamil with beta-blockers can increase the risk of bradycardia.
  - Beta-blockers including metoprolol, may exacerbate the rebound hypertension that can follow the withdrawal of clonidine.
  - Alcohol interferes with the extended release properties of this product.



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## Case Studies

- F, 55, femoral blood
  - Bzotropine - QUAL
  - Cyclobenzaprine – 58 ng/mL
  - Bupropion (w/ metabolite) – 75 ng/mL (670 ng/mL)
  - Tramadol (w/metabolite) – 830 ng/mL (ODM- 83 ng/mL)
  - Topiramate – 5400 ng/mL
  - Caffeine - QUAL
  - Cotinine - QUAL
  - Fentanyl (w/metabolite) – 1.1 ng/mL (NF 1.2 ng/mL)
  - Paroxetine – 63 ng/mL
  - Pregabalin – 7.7 mg/L
  - Ziprasidone – 13 ng/mL



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### Case Studies

- F, 55, femoral blood
  - Diphenhydramine – 0.328 mg/L
  - Dextromethorphan – 0.686 mg/L
  - Chlorpromazine – 0.076 mg/L
  - Trazodone – 0.399 mg/L
  - Olanzapine – QUAL
  - Donepezil – QUAL




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### Case Studies

- M, 56, femoral blood
  - Dextromethorphan – QUAL
  - Doxepin (w/metabolite) – 0.097 mg/L
  - Ketamine (w/metabolite) – 0.826 mg/L
  - Mirtazapine – QUAL
  - Cocaine (w/metabolites) – 38 ng/mL (heart blood, BE >800 ng/mL)
  - Nicotine and cotinine – QUAL
  - Caffeine – QUAL
  - Oxycodone – 23 ng/mL
  - THC (w/metabolites) – 1.9 ng/mL (heart blood)




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### Case Studies

- M, 32, femoral blood
  - Methadone (w/EDDP) – 0.672 mg/L
  - Mirtazapine – 0.101 mg/L
  - Nicotine and cotinine – QUAL
  - Caffeine – QUAL
  - Gabapentin – 3.8 mg/L
  - Hydroxyzine – 42 ng/mL




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### Case Studies

- F, 47, femoral blood
  - Gabapentin – 17 mg/L
  - Amitriptyline (w/ NT) – QUAL (0.342 mg/L)
  - Sertraline (w/ metabolite) – QUAL
  - Diphenhydramine – 0.520 mg/L
  - Lidocaine (w/ metabolite) – QUAL
  - Cotinine – QUAL
  - Cocaine (w/ metabolites) – 103 ng/mL (BE 128 ng/mL)
  - Urine: Also promethazine, dextromethorphan, doxylamine



### Case Studies

- F, 79, femoral blood
  - Amitriptyline (w/ NT) – 0.402 mg/L (0.415 mg/L)
  - Bupropion (with metabolite) – QUAL
  - Citalopram (w/ metabolite) – 0.466 mg/L
  - Diphenhydramine – 0.162 mg/L
  - Cotinine – QUAL
  - Oxycodone – 112 ng/mL
  - 7-aminoclonazepam – QUAL
  - Memantine - QUAL



### Case Studies

- M, 38, femoral blood
  - Clonazepam (w/ metabolite) – 2.7 ng/mL (44 ng/mL)
  - Clomipramine (w/metabolite) – 900 mg/L
  - Fluoxetine (w/metabolite) – 410 ng/mL (270 ng/mL)
  - Trazodone (w/metabolite) – 32 mcg/mL (0.059 mcg/mL)
  - Dextromethorphan (w/ metabolite) – 110 ng/mL (15 ng/mL)
  - Caffeine – QUAL
  - Cotinine – QUAL



### Final thoughts

- Polypharmacy may be necessary – but needs to be managed well
- Contraindicated, serious and significant risks of DIs may contribute to, but not be obvious in, fatal outcomes
- Patients and HCPs both share responsibility for appropriate polytherapy
- Regulatory references are living documents that reflect key DI information
- Prescribed, licit, and illicit drug combinations are the norm in death investigations – toxicologists can help wade through the information



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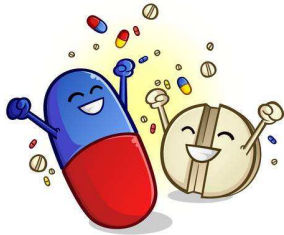
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### Questions



Luigino Apollonio  
lapollonio@cuyahogacounty.us

Thank you!



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